

Atomization and Combustion in LOX/H₂- and LOX/CH₄- Spray Flames Spray

M. Oswald¹, F. Cuoco², B. Yang³, M. De Rosa¹

¹German Aerospace Center (DLR), Lampoldshausen, Germany

²Avio S.p.A., Italy

³Northwestern Polytechnical University, China

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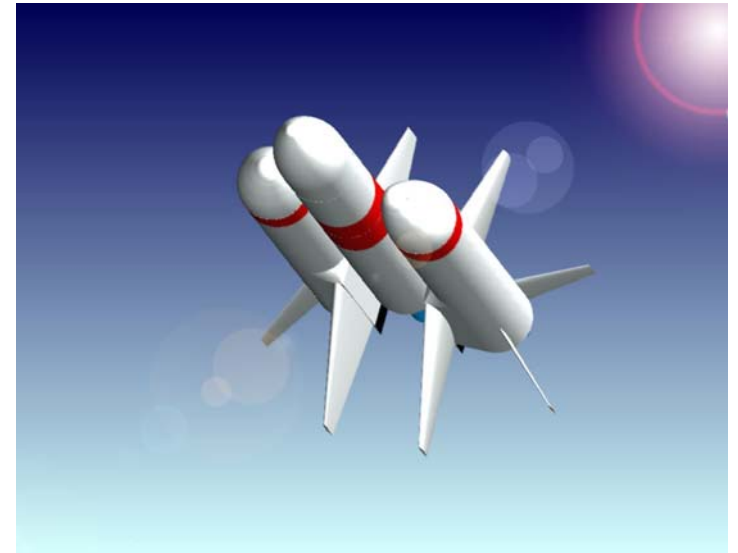
Motivation and Background

LOX/hydrocarbon: promising propellant combination for

- ▶ **high power booster engine (LFBB, RLV)**
- ▶ **upper stage**

advantages

- ▶ **low costs**
- ▶ **easy ground operation**
- ▶ **high performance (I_{sp} lower than H_2 , thrust/weight ratio higher than H_2)**
- ▶ **low toxic potential (green propellant)**



HC candidates for booster engines: kerosene, methane

trade-off between kerosene and CH₄

- ▶ I_{sp} =thrust/mass of propellant
- ▶ thrust/weight
- ▶ tank masses
- ▶ chamber cooling:
 - cooling capability
 - pressure drop
 - coking behavior
- ▶ combustion:
 - soot formation
 - combustion stability

CH₄ for basic investigations of LOX/HC-combustion

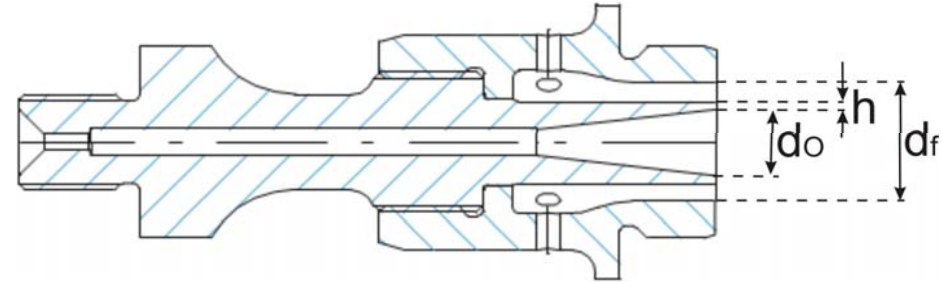
- ▶ simple kinetics
- ▶ well defined composition

as compared to kerosene

trade-off under discussion

Why to compare LOX/H₂ with LOX/CH₄?

- ▶ lot of data on LOX/H₂ spray combustion available
- ▶ LOX/CH₄ and LOX/H₂ use coaxial injectors
 - CH₄ injected at typ. 280K
 - LOX injected at typ. 120K
 - $v_{CH_4} \gg v_{LOX}$



propellant injectors are key components

- ▶ combustion efficiency
- ▶ combustion stability
- ▶ thermal and chemical load on combustor walls

Rahman S.A., Santoro R.J., "A Review of coaxial gas/liquid spray experiments and correlations", AIAA 94-2772, 1994

Vingert L., Gicquel P., Lourme D., M  noret L., "Coaxial injector atomization", in AIAA Progress in Astronautics and Aeronautics, Vol. 169, 1995

relevant thermo-physical properties of O₂, CH₄, H₂

	O ₂	CH ₄	H ₂	
critical temperature	154.6	190.5	32.9	[K]
critical pressure	5.04	4.60	1.28	[MPa]
reduced pressure P/P_{crit}^1	1.19	1.30	4.69	
reduced pressure T/T_{crit}^1	0.65	1.47	3.65	
density ¹		47.3	11.7	[kg/m ³]
viscosity ¹		12.0	4.94	[μPa·s]
specific heat ¹		43.89	32.3	[J/mol·K]
thermal conductivity ¹		0.038	0.11	[W/m·K]
laminar flame velocity ¹		3.93	10.7	[m/s]
ignitability limits		5.1-61	4-94	[Vol %]

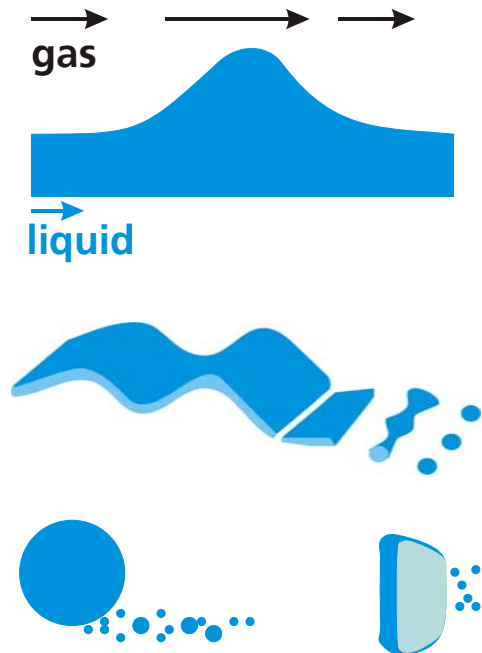
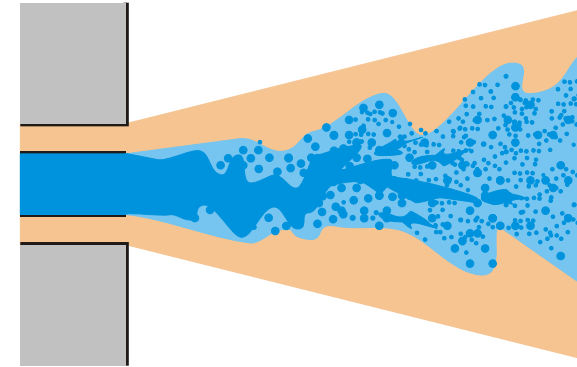
¹at injector exit conditions: $P_C = 6 \text{ MPa}$, $T_{\text{H}_2} = 120 \text{ K}$, $T_{\text{CH}_4} = 280 \text{ K}$

Atomization

complex interaction of several forces

- ▶ aerodynamic forces
- ▶ surface tension
- ▶ viscosity
- ▶ turbulence level of liquid jet and gas flow
- ▶ sudden change of boundary conditions at injector exit
- ▶ basic mechanisms leading to atomization neither completely identified nor well modeled

Ledoux M., Caré I., Micci M., Glogowski M., Vingert L., Gicquel P.,
Atomization of Coaxial Injectors, 2nd Int. Symp. on LRP, Chatillon, 1995



Injector scaling

scaling

- ▶ injector geometry

$$d_o, d_f, h, \dots$$

- ▶ flow conditions at injector exit

$$V_o, V_f, \rho_o, \rho_f, \mu_o \dots$$

- ▶ groups of non-dimensional numbers

Weber number $We = \frac{\rho_{fuel} (V_{fuel} - V_{ox})^2 d_{ox}}{\sigma}$

momentum flux ratio $J = \frac{(\rho V^2)_{fuel}}{(\rho V^2)_{ox}}$

liquid Reynolds number $Re_l = \frac{\rho_{ox} V_{ox} d_{ox}}{\mu_{ox}}$

velocity ratio $R_V = \frac{V_{fuel}}{V_{ox}}$

Ohnesorge number $Oh = \frac{\sqrt{We}}{Re}$

Injector scaling: examples

atomization regime

(Farago, Chigier):

$$We < 25$$

$$25 < We < 100$$

$$100 < We < 500$$

Rayleigh breakup

membrane-type breakup

fibre-type breakup

droplet size

(Rahman, Santoro)

$$D \propto d_o^a \sigma^b$$

$$0 \leq a \leq 2.9$$

$$-0.2 \leq b \leq 0.3$$

intact core length

(Villiermaux)

$$x \approx \frac{6}{\sqrt{J}}$$

problem: correlations derived under non-representative conditions for rocket propulsion

- ▶ H₂O as substitute for LOX
- ▶ cold flow
- ▶ data for LOX/H₂ only for specific configurations

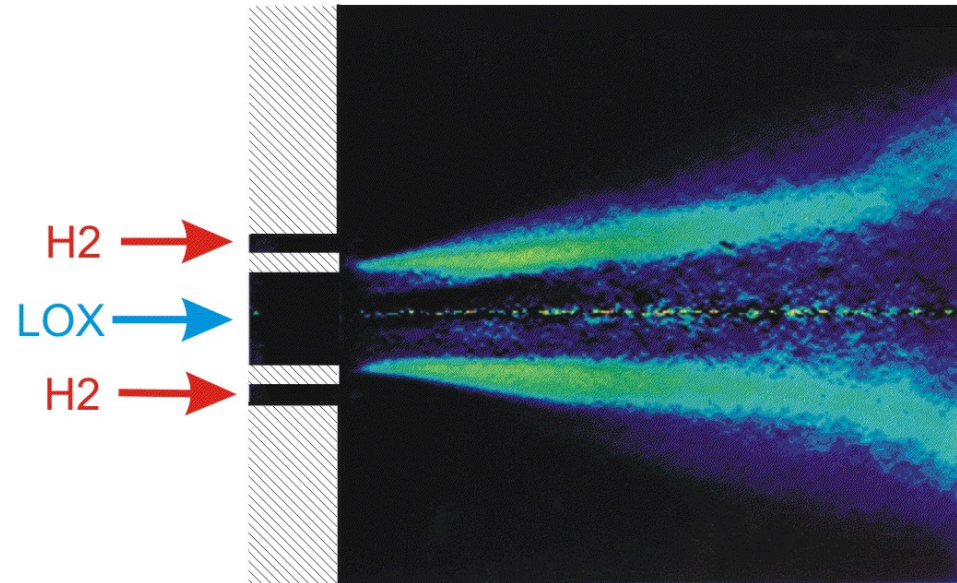
H₂O as substitute in cold flow tests

property	H ₂ O	LOX			
	1 bar	1 bar	10 bar	30 bar	100 bar
surface tension σ [mN/m]	73	15	7	3	supercritical state
viscosity μ [μ Pa·s]	1000	195	99	59	30

- ▶ difficult to adjust non-dimensional numbers including σ and μ with H₂O
- ▶ We, Re_{liq} can be more than an order of magnitude different from representative conditions

Atomization in hot fire conditions

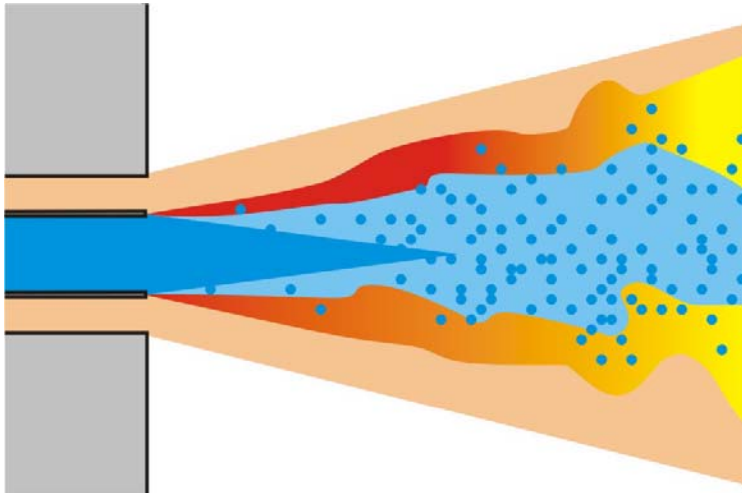
- ▶ flame between LOX-jet and annular fuel-flow
- ▶ influence of heat release and mixing layer of reactants/products on LOX-jet disintegration
- ▶ fluid properties at injector exit (We , J , Re_l , ...) sufficient to characterize atomization in hot fire tests?



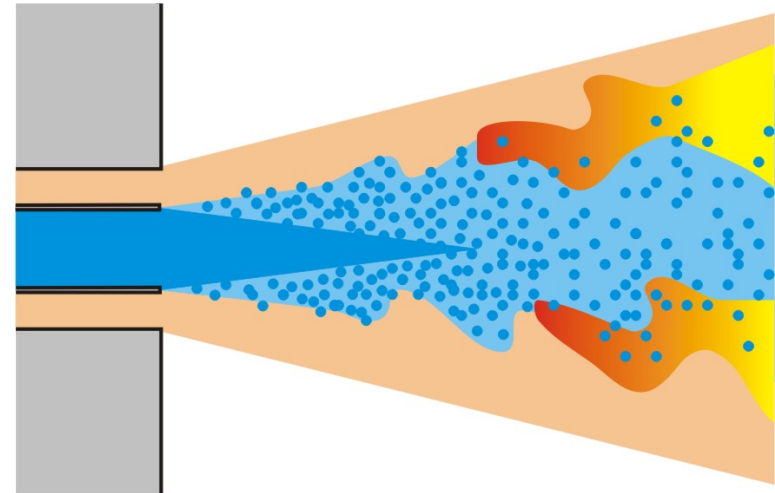
OH-imaging of flame at
 $P_c = 8$ MPa at P8 test
facility

flame stabilization mechanisms

flame anchored at LOX-post



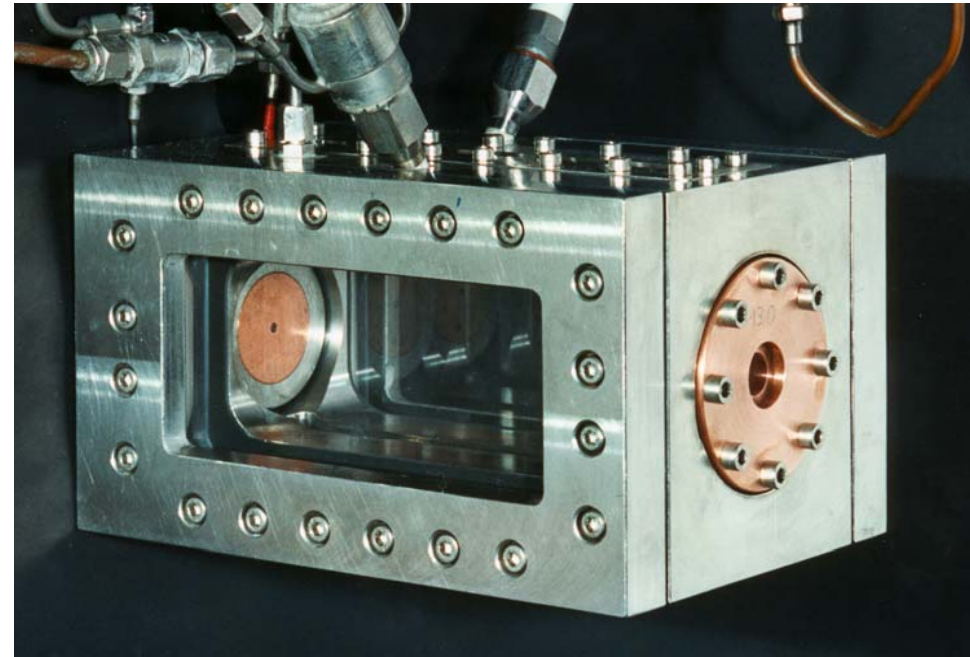
lifted flame



- ▶ influence of kinetic and thermo-physical properties of propellants on flame stabilization mechanism?
- ▶ influence of stabilization mechanism on atomization?

M3 test facility, micro combustor

- ▶ **propellants at representative conditions**
 - LOX, H₂ @ 88K
 - CH₄ @ 280K
- ▶ **L=14cm, A=6x6cm²**
- ▶ **single coaxial injector**
- ▶ **max. P_c=1.5 MPa**
- ▶ **max. run time 3s**
- ▶ **full optical access**
- ▶ **pressure representative for ignition transients**
- ▶ **hot fire tests!**



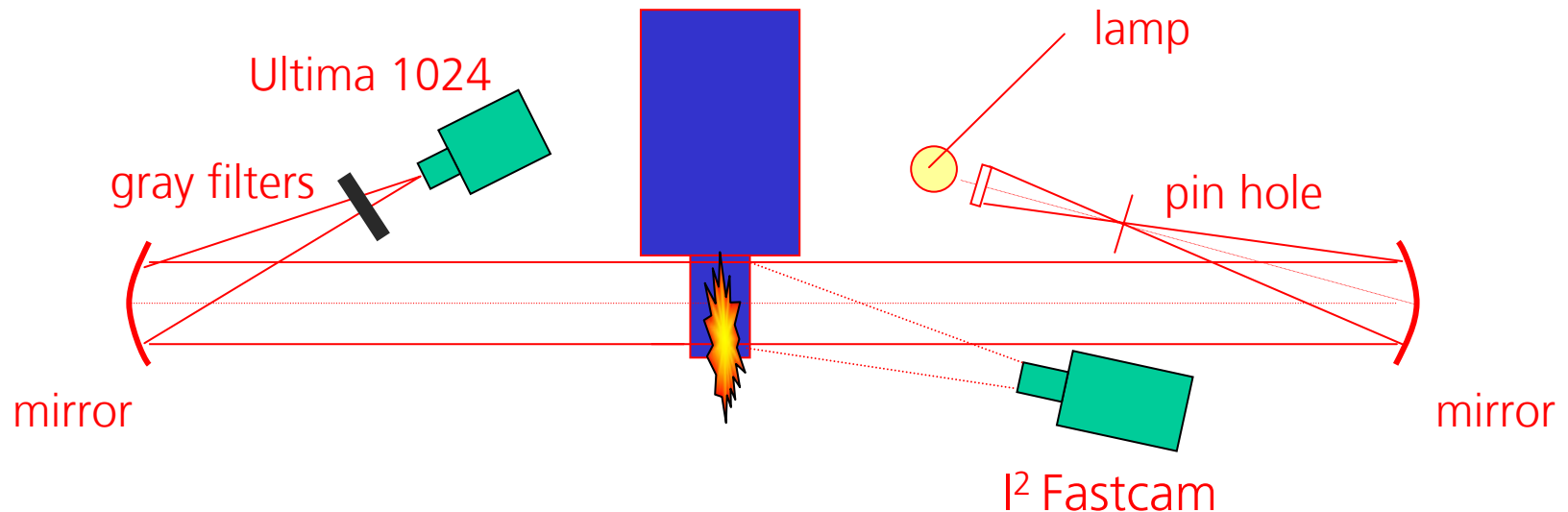
optical diagnostics

flame visualization

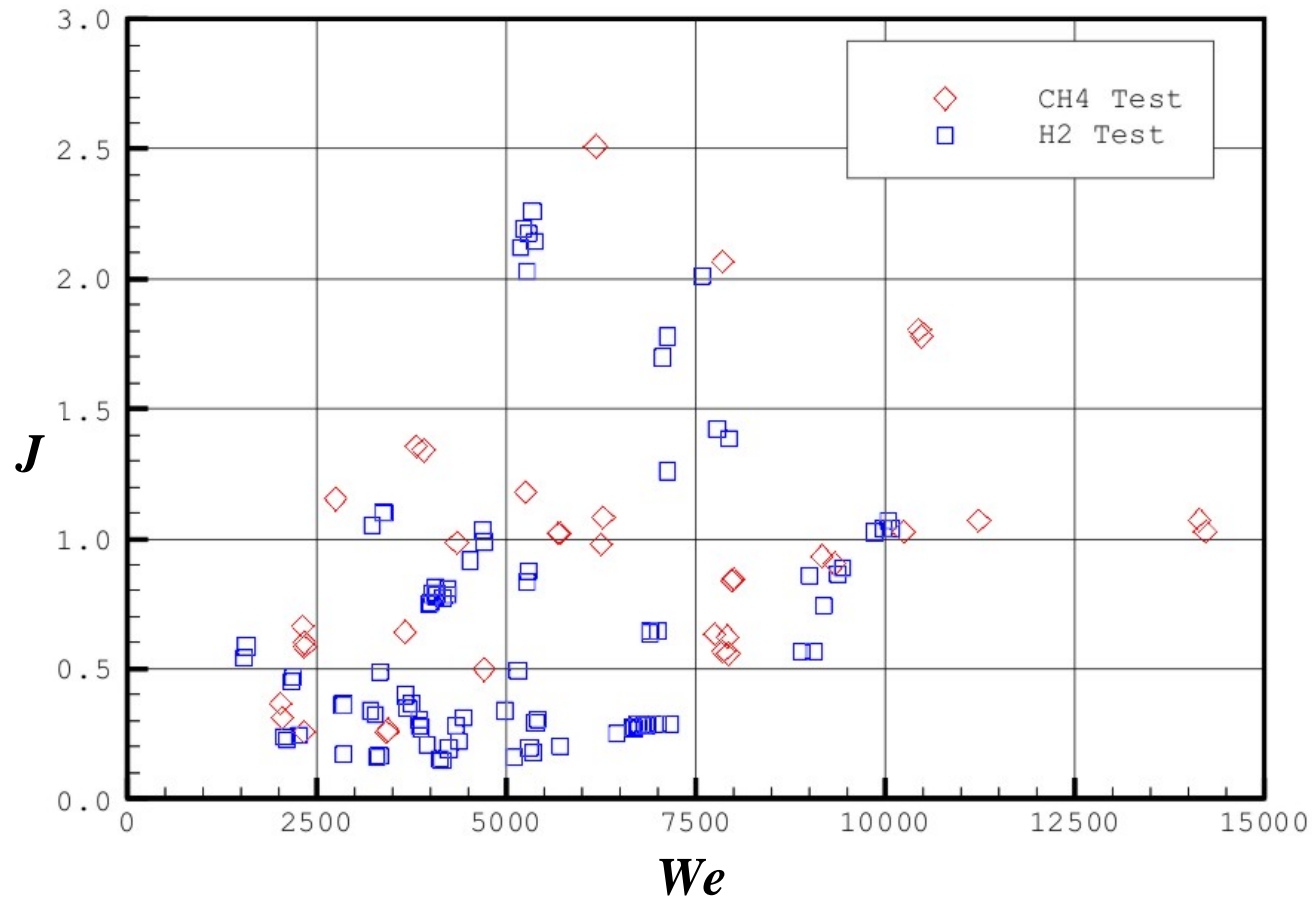
- ▶ high speed OH-imaging ($\approx 310\text{nm}$)
- ▶ ULTIMA I² ICCD
- ▶ up to 27 kfps

flow visualization / liquid phase

- ▶ Schlieren photography
- ▶ Kodak Flowmaster



Test conditions

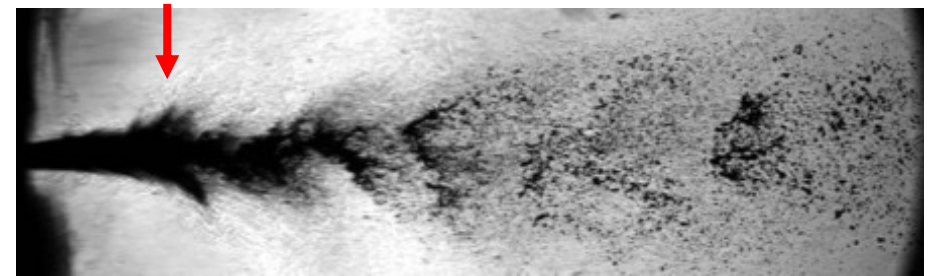


- independent variation of Weber-number We and momentum flux ratio J

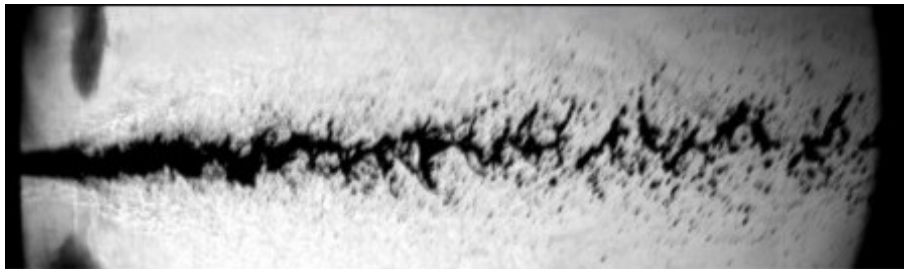
LOX spray pattern for CH_4/LOX spray flames ($P_c=1.5\text{bar}$)



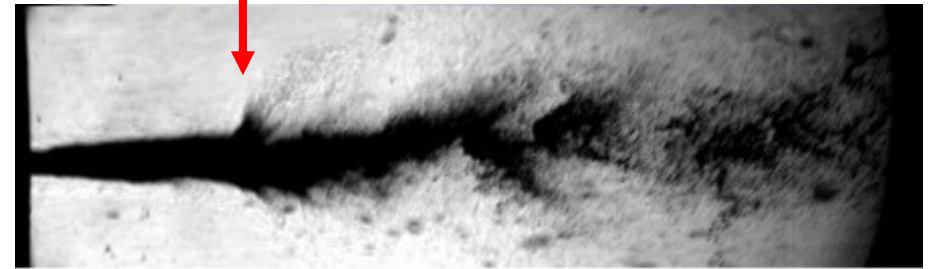
$We=3812, J=1.35$



$We=10450, J=1.8$



$We=2335, J=0.60$



$We=7936, J=0.56$

increasing J :

- ▶ higher dispersion of liquid phase
- ▶ decreasing visible intact core length

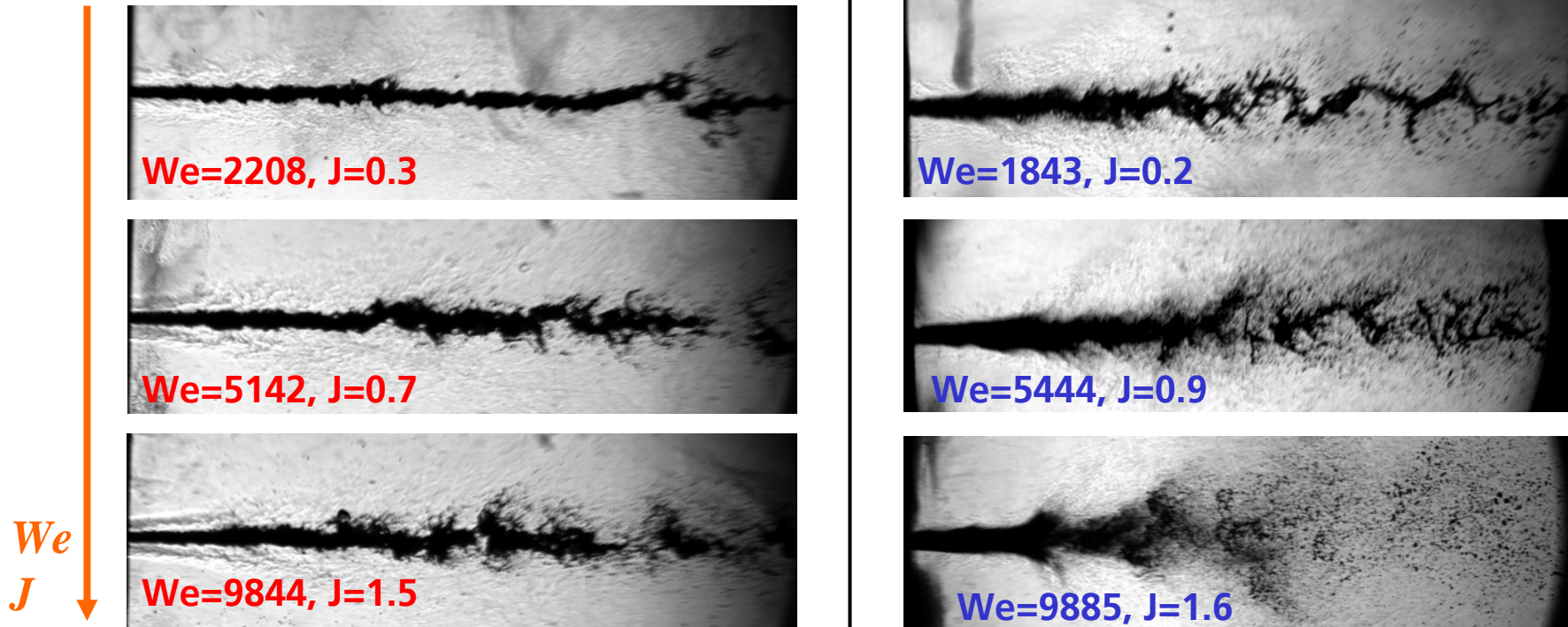
increasing We :

- ▶ smaller droplets
- ▶ sudden change of atomization behaviour

LOX-spray pattern for **LOX/H₂** and **LOX/CH₄**-spray flames

LOX/H₂

LOX/CH₄

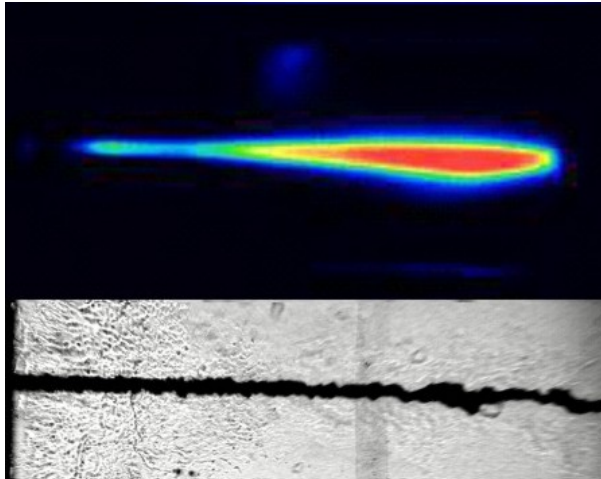


- ▶ similar trends for variation of We and J for both propellants
- ▶ atomization significantly more efficient for CH₄
- ▶ visible breakup length much larger for H₂ than for CH₄

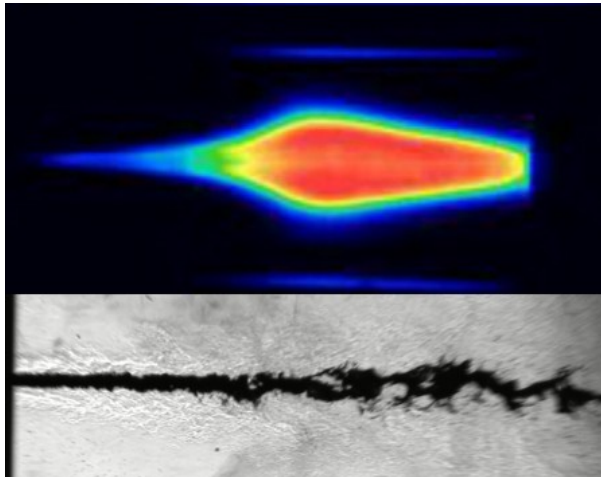
Flame and LOX-spray pattern for **LOX/H₂** and **LOX/CH₄**

LOX/H₂

We=2192
J=0.47

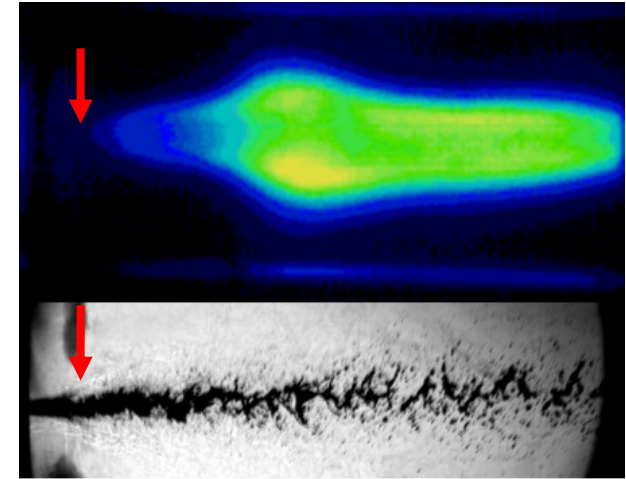


We=7007
J=0.65

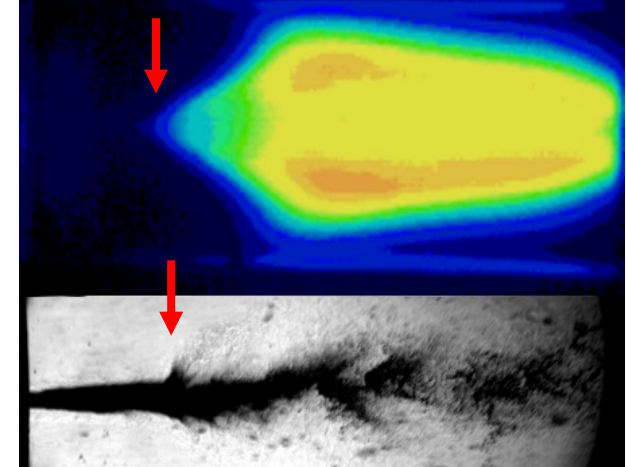


LOX/CH₄

We=2335
J=0.60



We=7936
J=0.56



- ▶ significantly larger flame spreading angle for CH₄
- ▶ anchored flames for H₂, lifted flames for CH₄

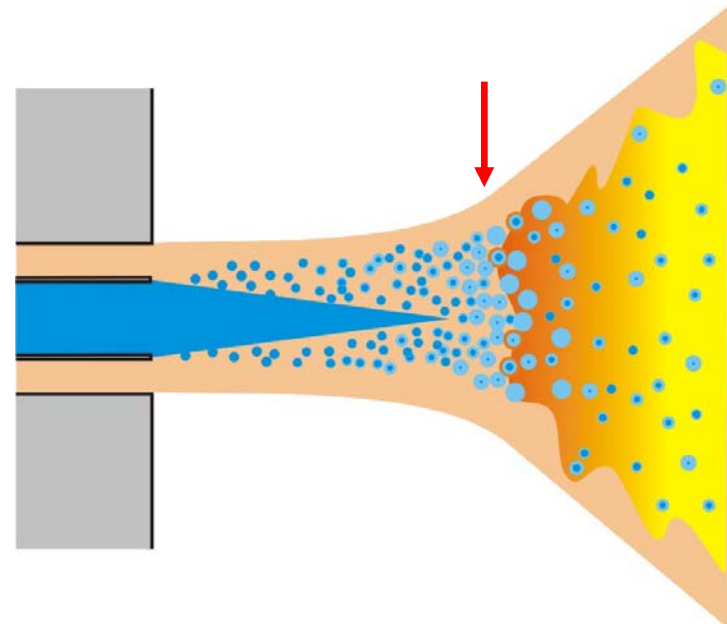
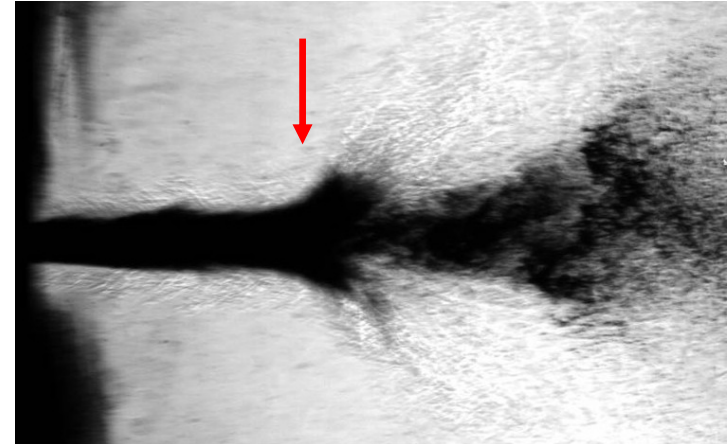
Stabilization of lifted flame in LOX/CH₄ spray

upstream the stabilization point:

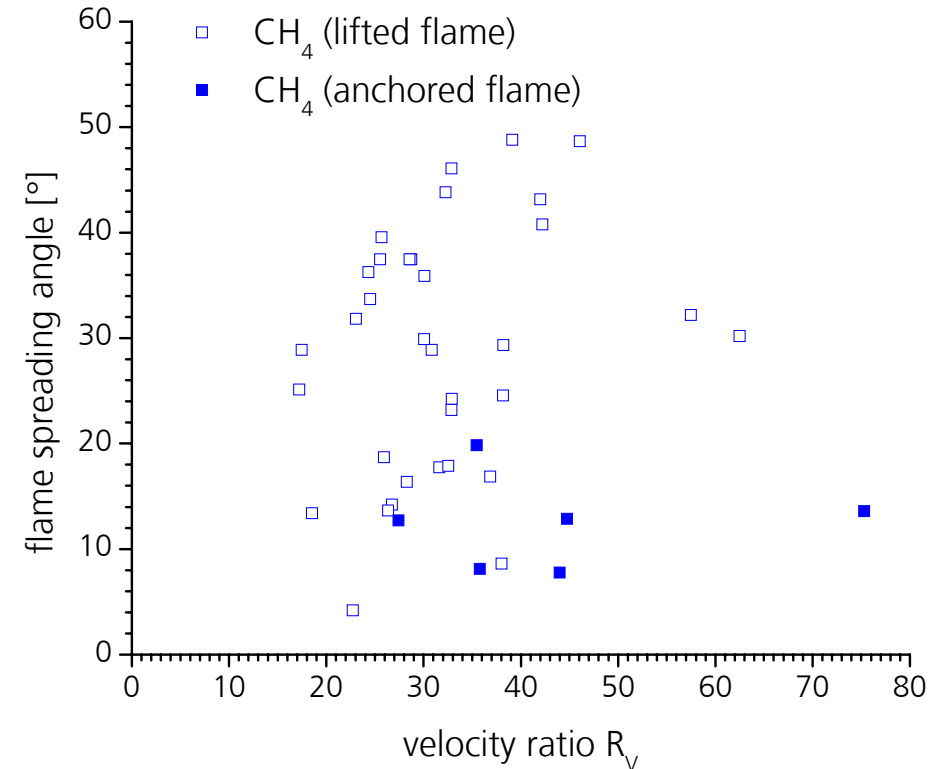
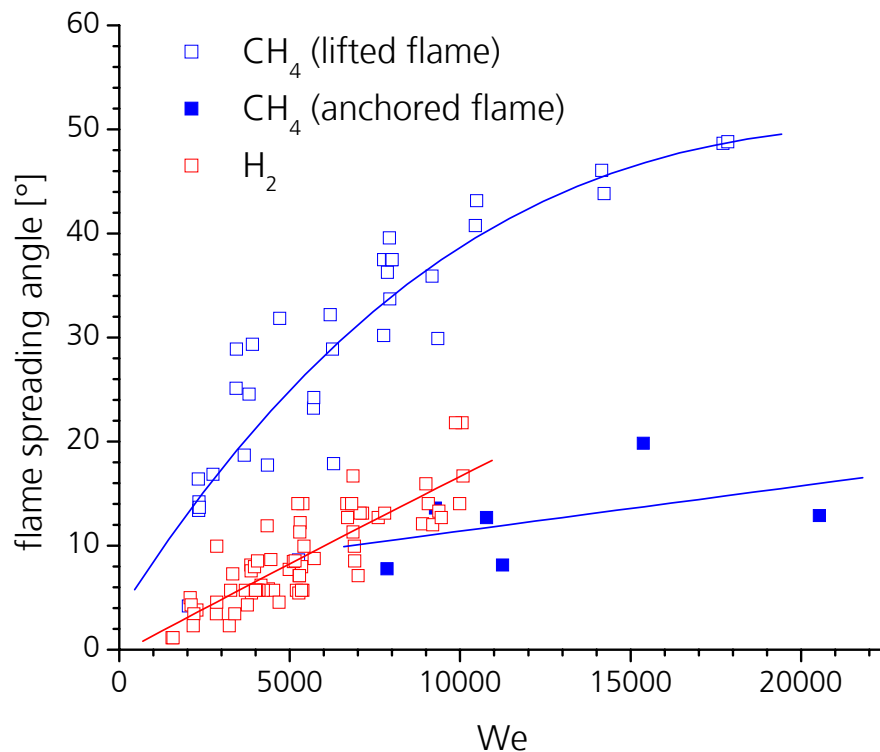
- ▶ atomization
- ▶ droplet evaporation
- ▶ mixing of CH₄ and GO₂
- ▶ increasing R_{OF} due to LOX evaporation

downstream the stabilization point:

- ▶ distributed reaction
- ▶ heat release, production of reactants
- ▶ flame position depends on amount of GO₂ / local mixture ratio

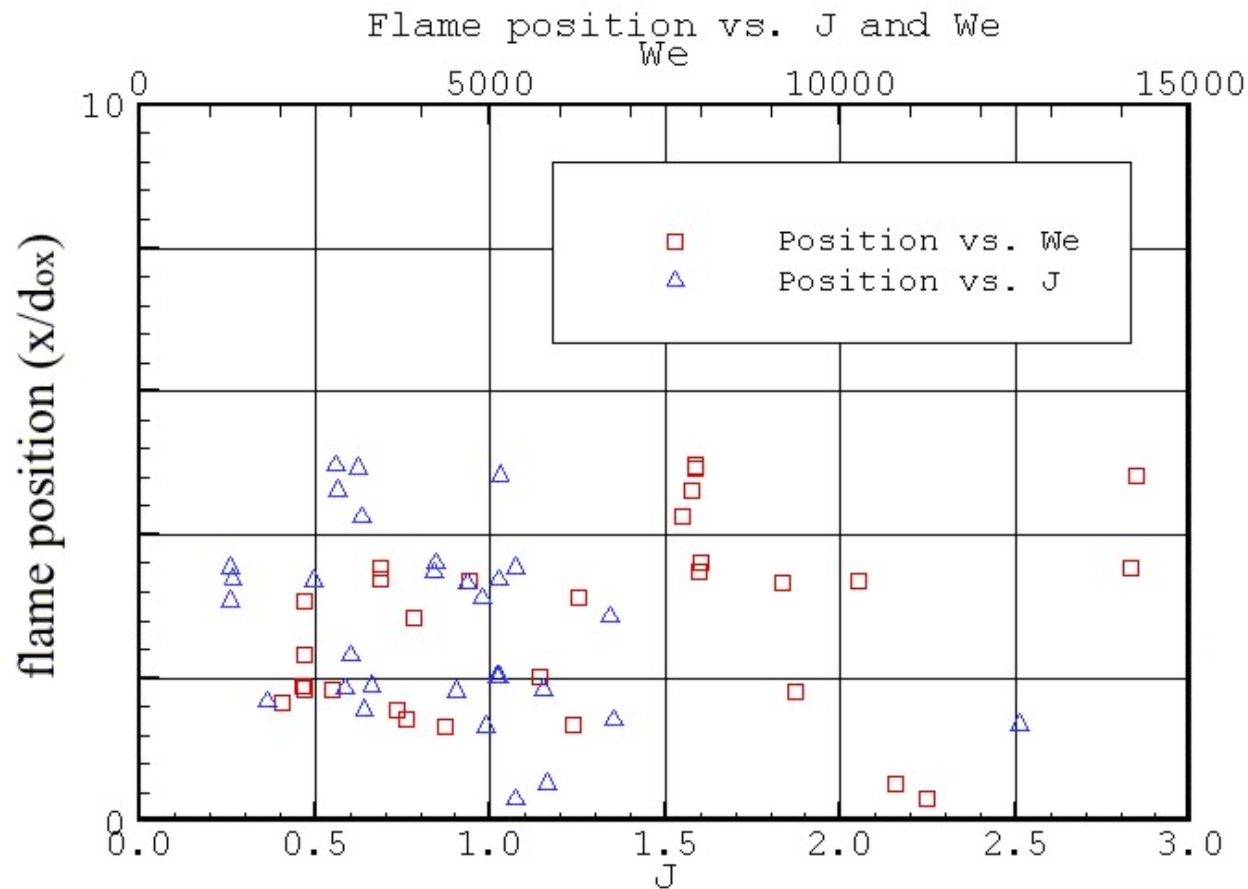


Flame spreading angle for LOX/CH₄- and LOX/H₂-spray flames



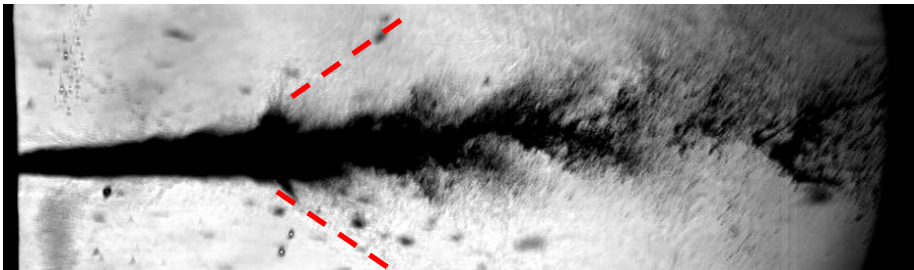
- ▶ best correlation with We
- ▶ no correlation with J , Oh , R_v , Re_{liq}
- ▶ large spreading angles for lifted flames

Lift-off distance as function of We and J

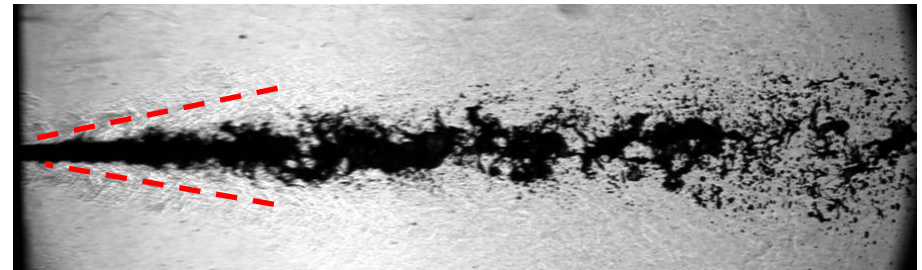


- no correlation of flame lift-off distance with any of the non-dimensional numbers J , Oh , R_v , Re_{liq} , We found

Effect of combustion chamber pressure on atomization and flame pattern for LOX/CH₄-spray flame



$P_c = 1.5 \text{ bar}$
 $We = 7260, J = 0.5$



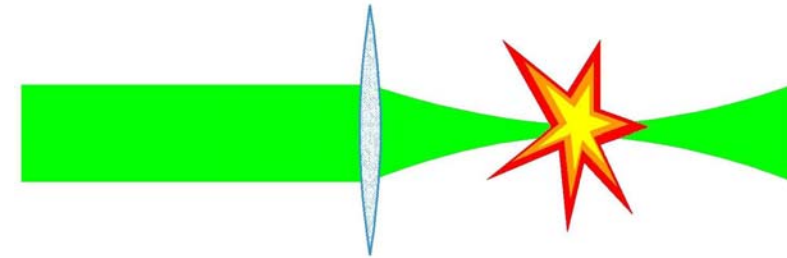
$P_c = 3.0 \text{ bar}$
 $We = 8417, J = 0.5$

- ▶ lifted flames at pressures above 3bar
- ▶ for lifted flames significantly more violent atomization process downstream the flame anchoring position

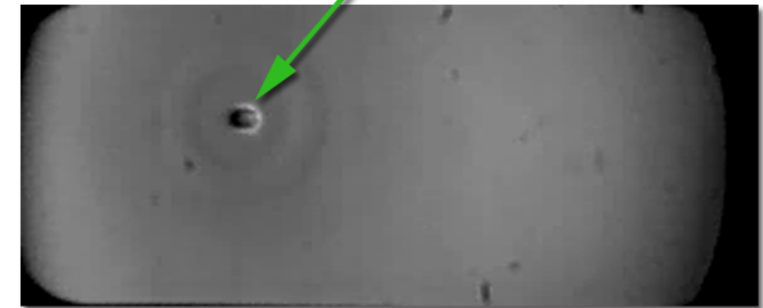
Ignition transient

ignition by laser induced gas break down

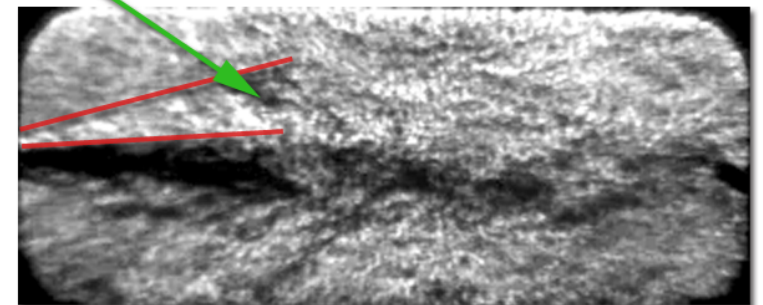
- ▶ full control of time and location of ignition
- ▶ no distortion of the flow due to ignition equipment
- ▶ energy deposition
 - Nd:YAG-laser, 532nm , 195 mJ/pulse, 10ns : $\sim 10\text{GW}/\text{cm}^2$ in the laser focus
 - focus-position:
 - $z = 36\text{ mm}$ downstream injector
 - $r=2.5\text{mm}$ off-axis
 - results independent on laser pulse energy (80-195mJ/pulse)



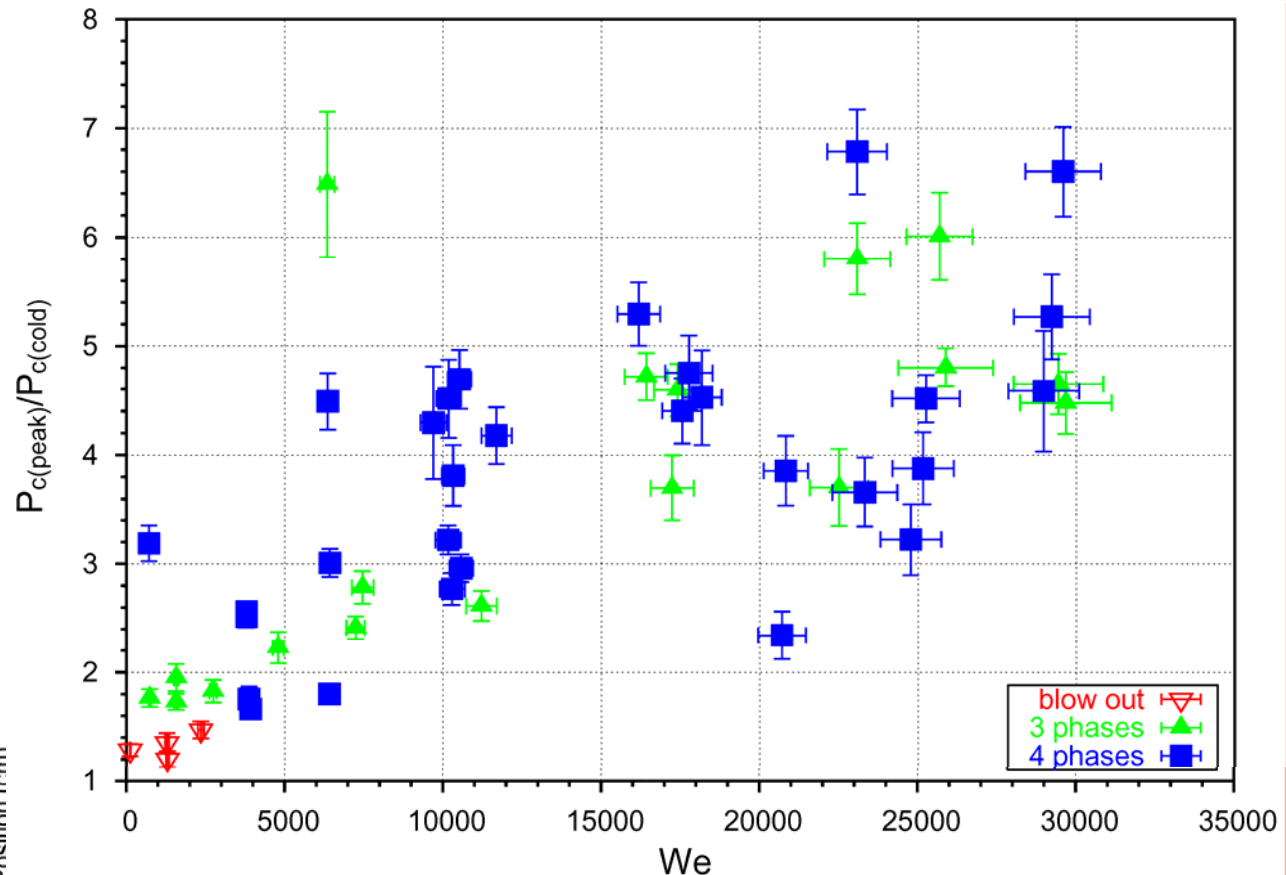
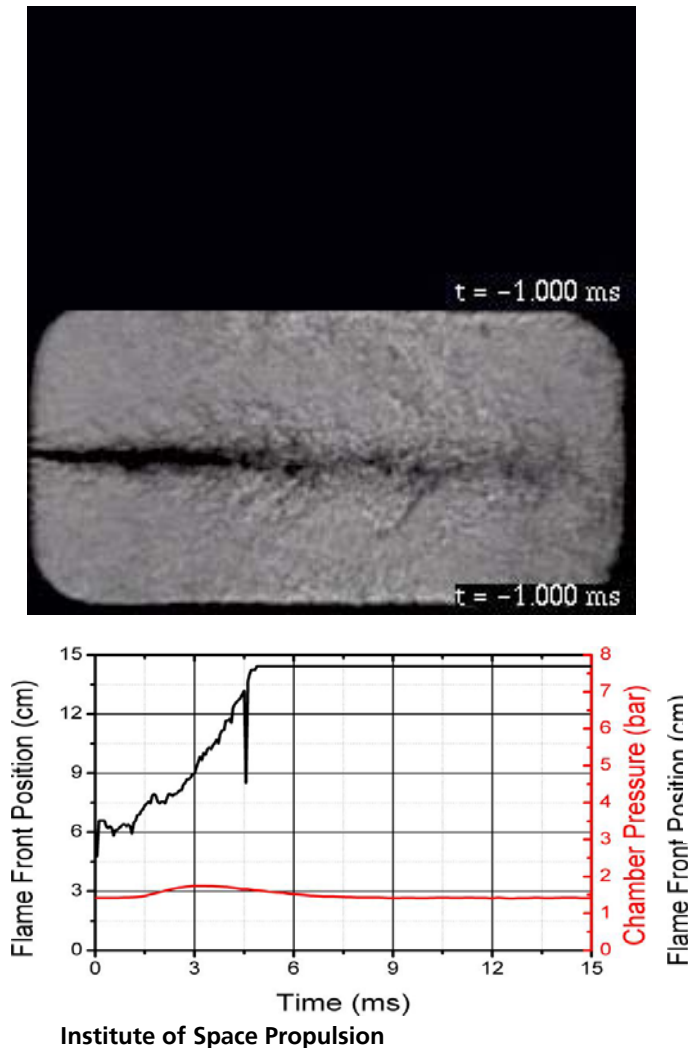
Laser induced plasma



Laser induced flame kernel

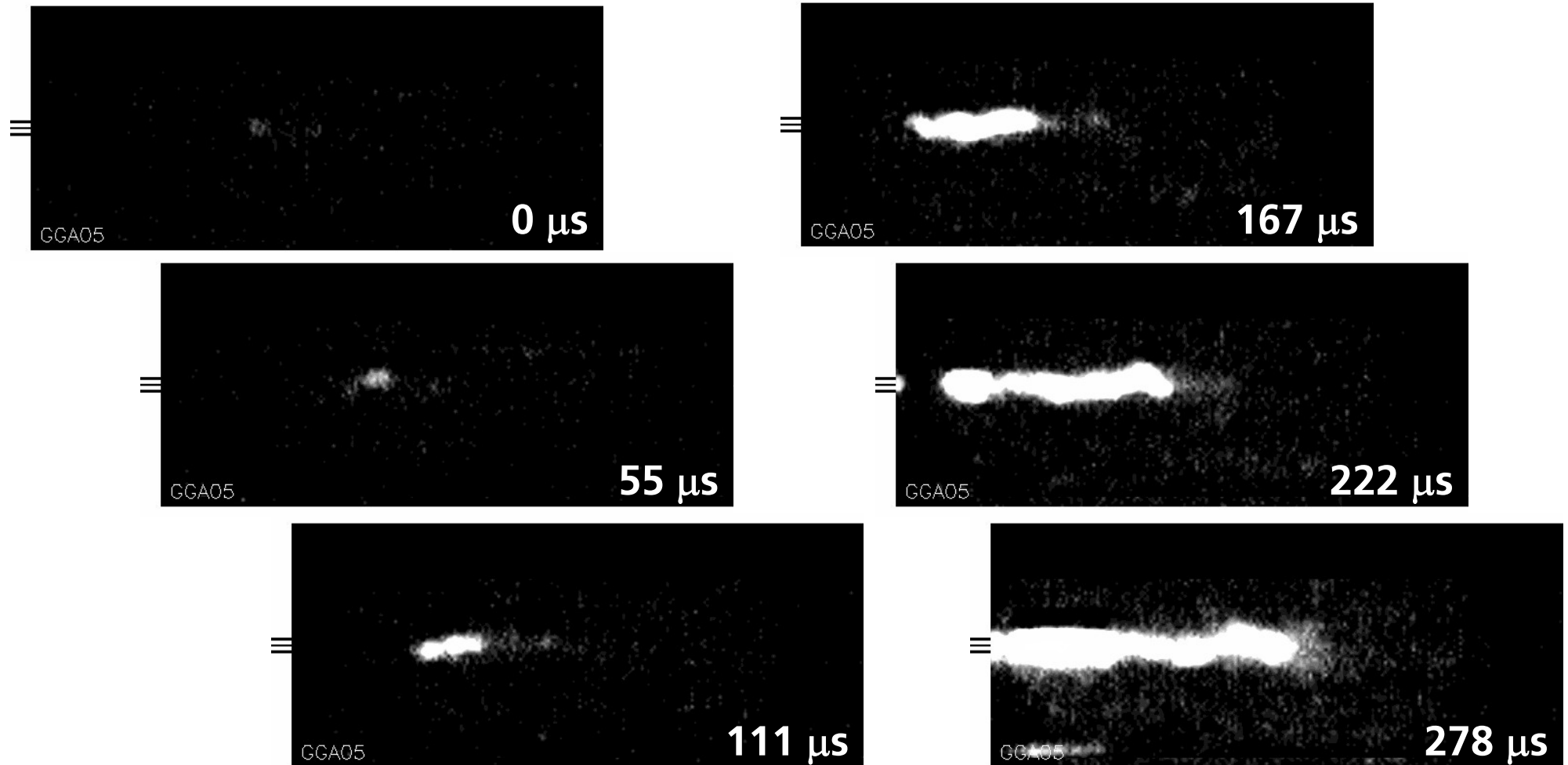


flame evolution for the 3 types of ignition scenarii (LOX/H₂):

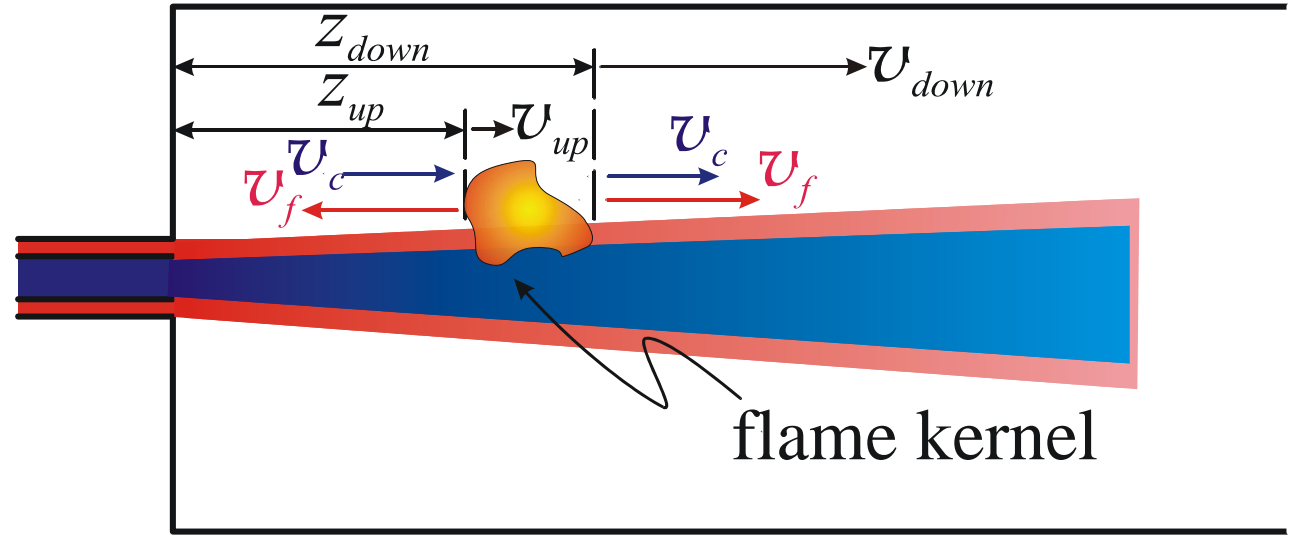
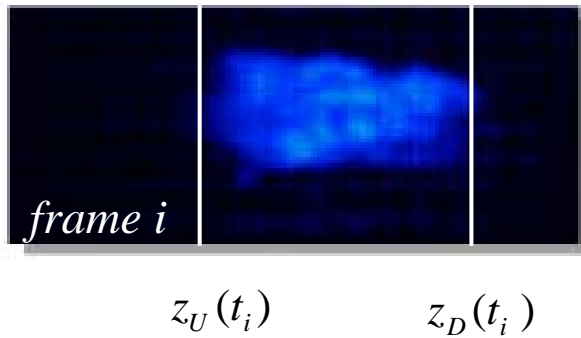


- ▶ ignition pressure peak well correlated with We
- ▶ flame blow-out clearly correlated with H₂-momentum flow $I_{H2} = \rho v^2 A$: blow-out for $I_{H2} < 0.8 \text{ kg} \cdot \text{m/s}^2$

flame kernel evolution



data reduction



flame front velocities

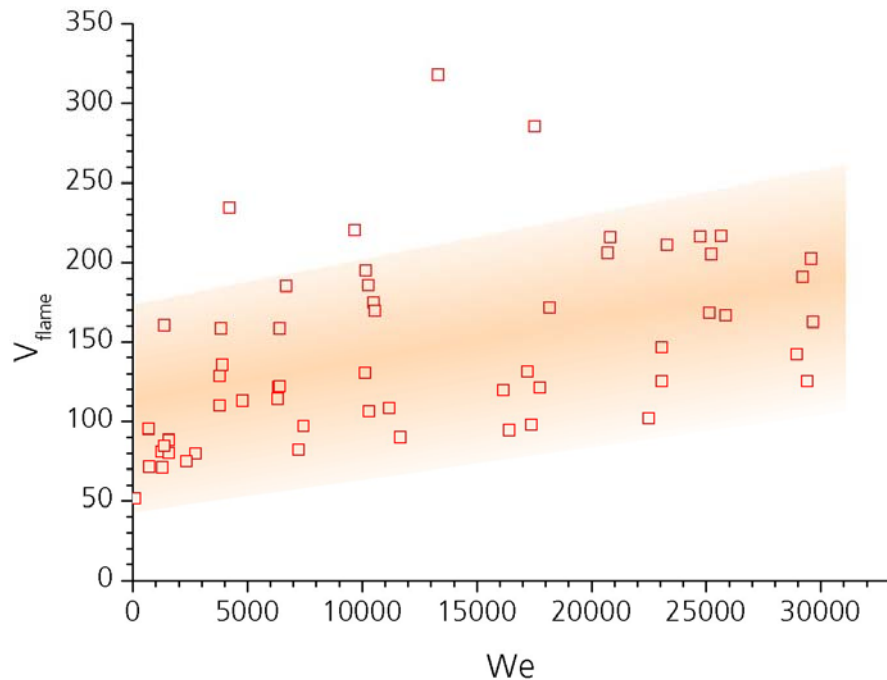
- ▶ determination of upstream- and downstream flame front positions: $z_U(t_i)$, $z_D(t_i)$
- ▶ determination of upstream- and downstream flame front velocities: $v_U = \frac{z_U(t_{i+1}) - z_U(t_i)}{\Delta t}$

$$v_D(z_D) = v_C(z_D) + v_F(z_D) \quad v_U(z_U) = v_C(z_U) - v_F(z_U)$$

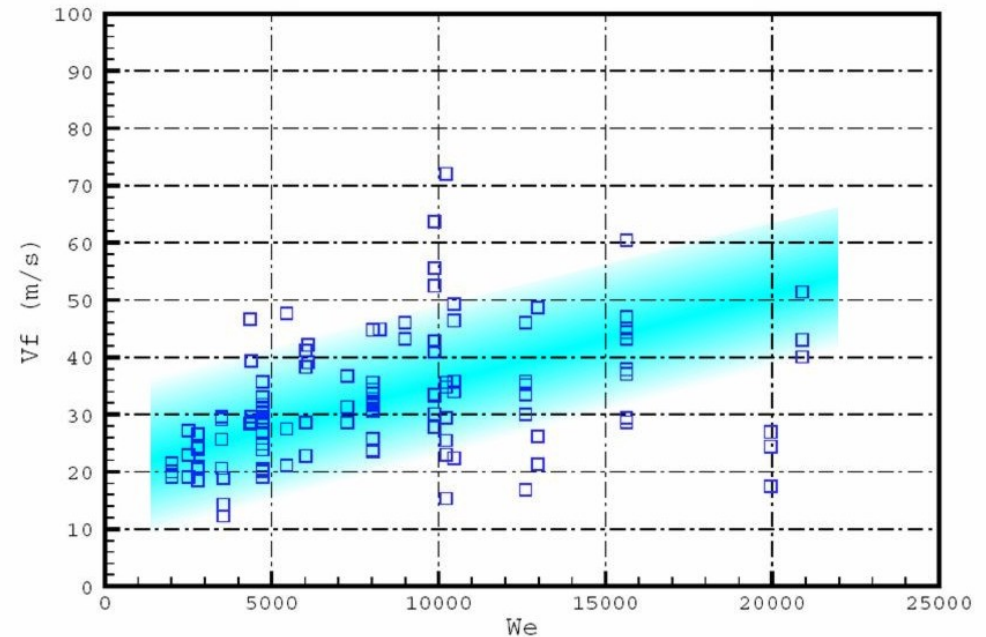
- ▶ determination of flame front- and convection velocities: $v_C = \frac{v_D + v_U}{2}$ $v_F = \frac{v_D - v_U}{2}$

flame front velocities

LOX/H₂



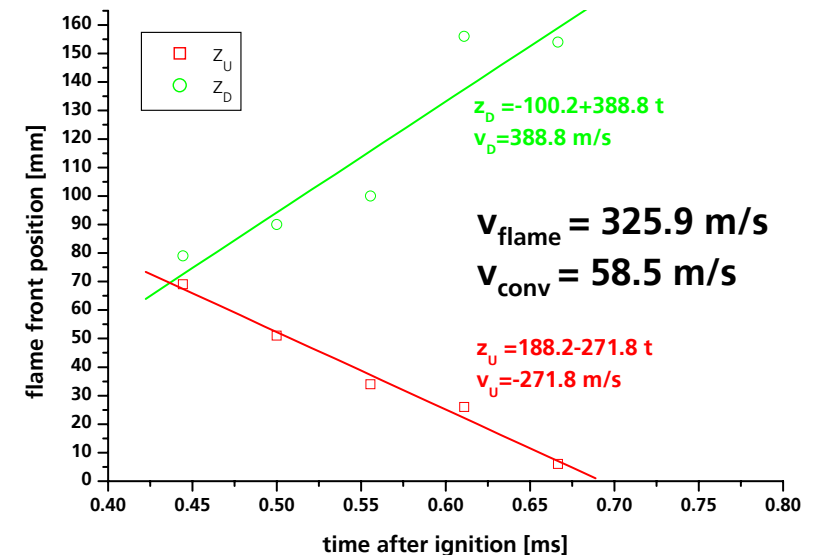
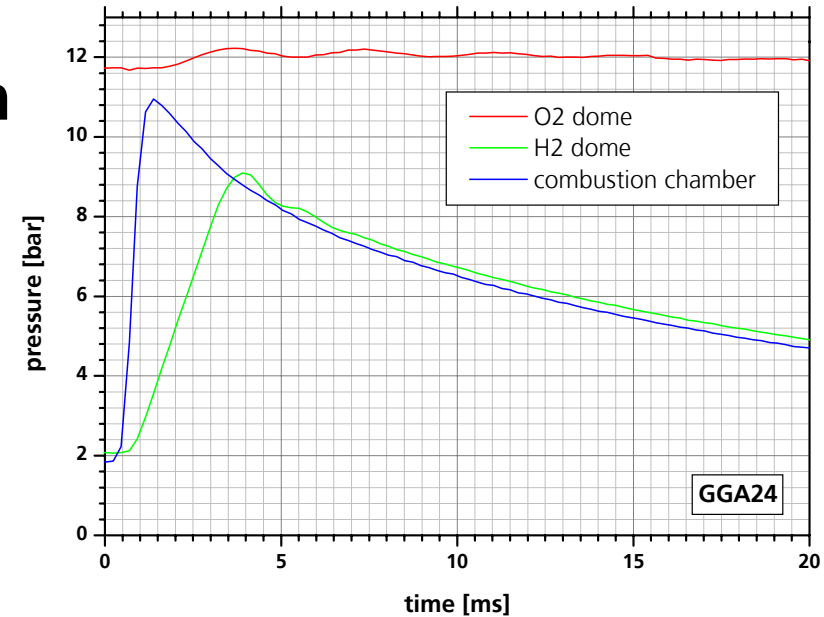
LOX/CH₄



- ▶ flame front velocity \gg laminar burning velocity
- ▶ weak correlation with We , no strong correlation with J , Oh , R_v , Re_{liq}
- ▶ $(v_{\text{flame}})_{\text{H}_2} / (v_{\text{flame}})_{\text{CH}_4} \approx 3 - 5$; ratio of laminar burning velocities 2.7

ignition tests for code validation

- ▶ injection of GO_2/GH_2 to reduce complexity
- ▶ determination of
 - pressure evolution during ignition transient
 - convective velocity of flame kernel
 - flame front velocity of expanding flame kernel
- ▶ data used for code validation by
 - ONERA Châtillon
 - SNECMA Vernon
 - CERFACS Toulouse
 - DLR Lampoldshausen



Coaxial injection at supercritical pressure

Binary liquid N₂/gaseous He system

$d_{\text{LN}_2} = 1.9 \text{ mm}$, $v_{\text{LN}_2} = 5 \text{ m/s}$, $v_{\text{He}} = 100 \text{ m/s}$, $T_{\text{LN}_2} = 97 \text{ K}$, $T_{\text{He}} = 280 \text{ K}$.

A: $P_c = 1.0 \text{ MPa}$, subcritical N₂



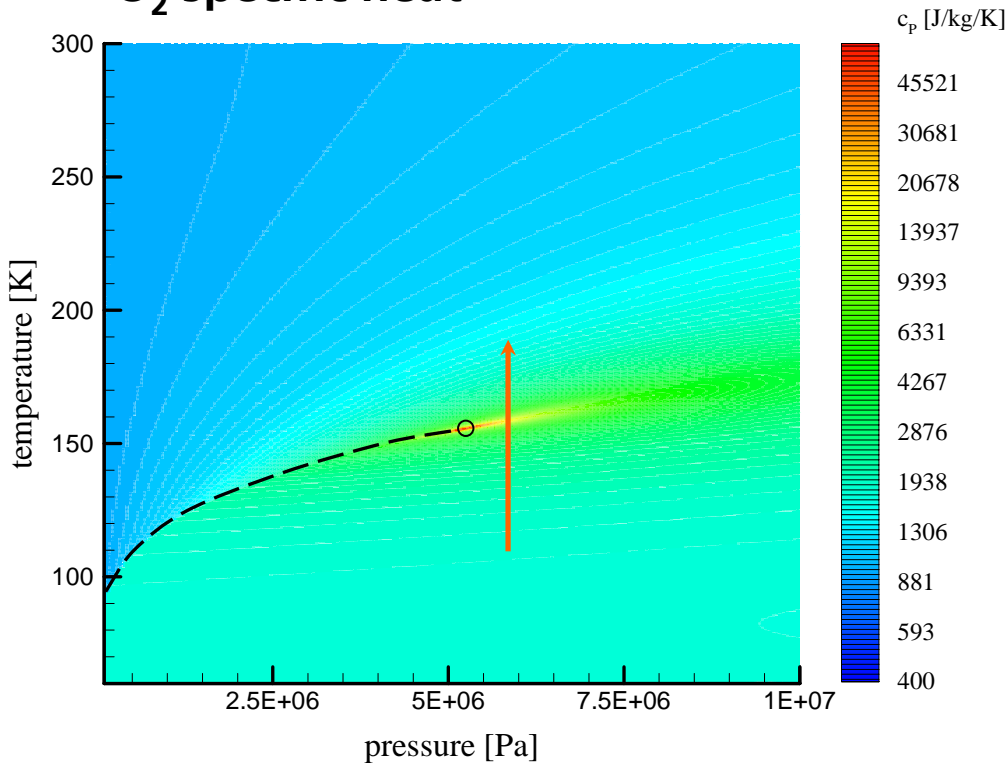
B: $P_c = 6.0 \text{ MPa}$, transcritical N₂

- ▶ spray formation at subcritical pressure
- ▶ reduced surface tension approaching the critical point
- ▶ turbulent mixing of dense and light fluid components at at supercritical pressure

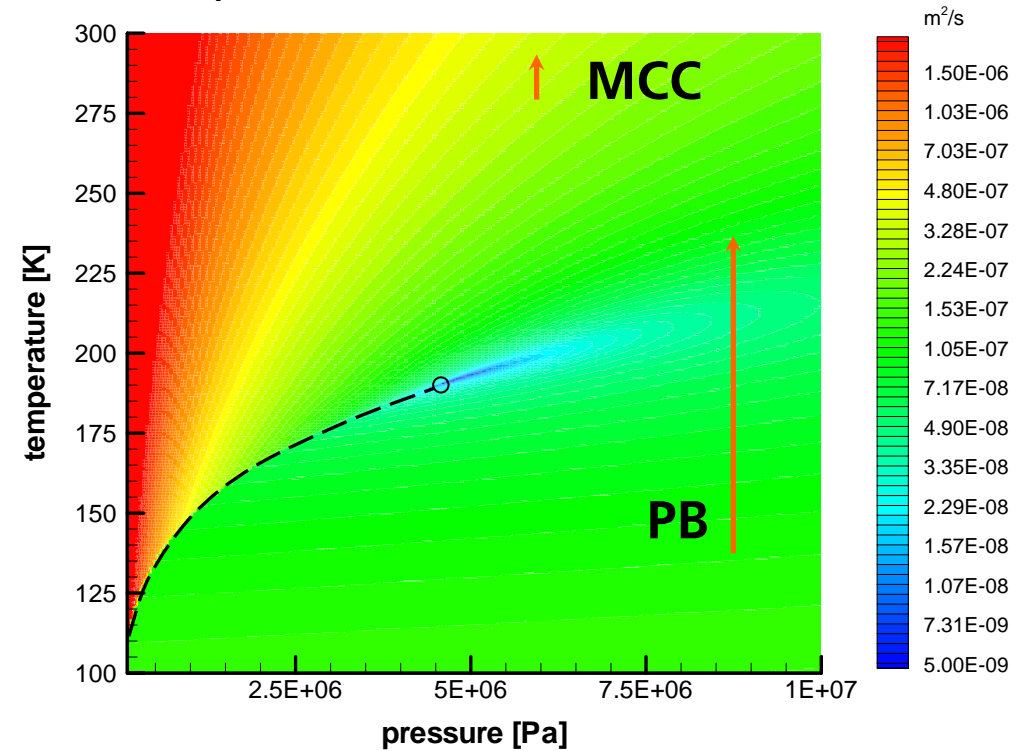


Thermo-physical properties in the near critical region

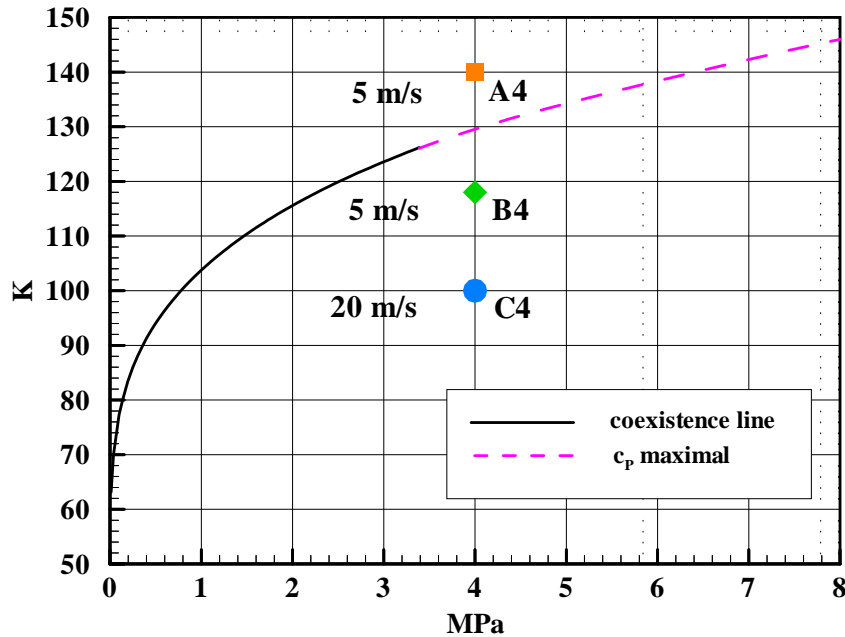
O₂ specific heat



CH₄ thermal diffusivity



LN₂ free jet-decay, Raman scattering

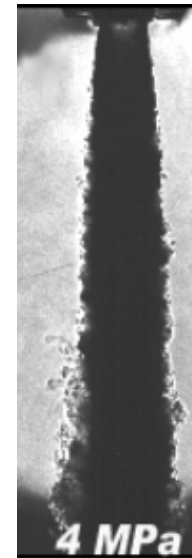


Injection temperatures:

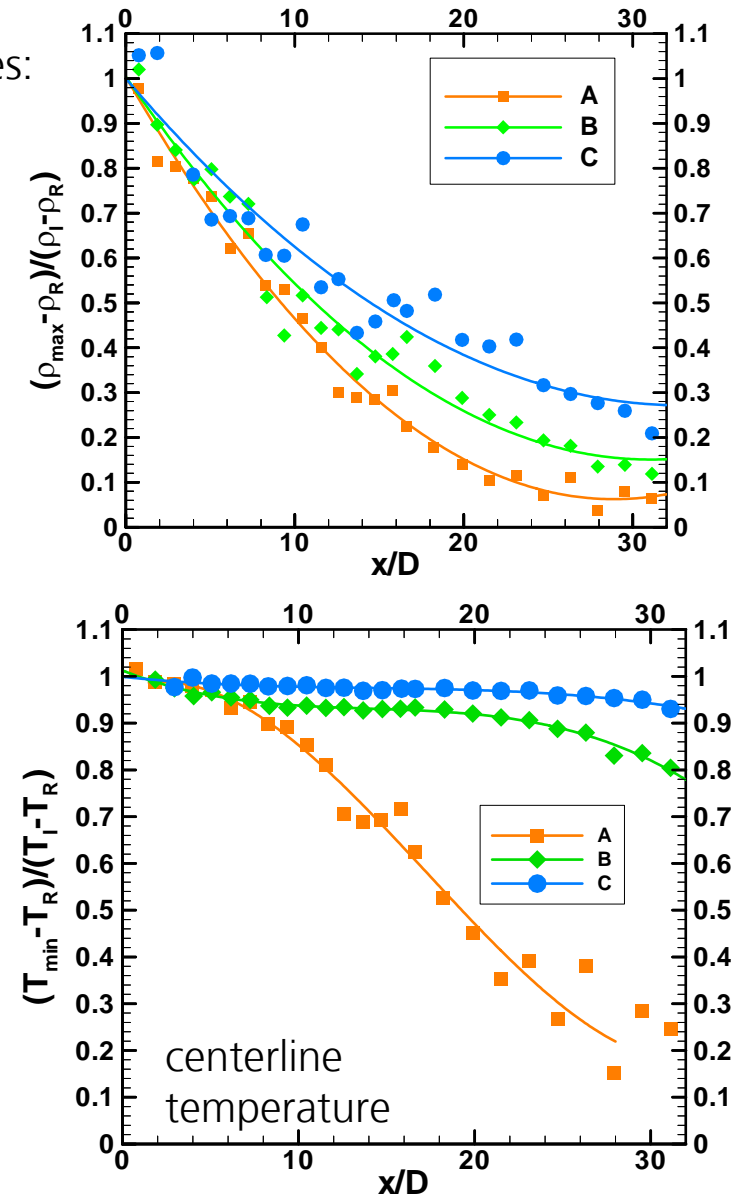
A4: 140K

B4: 118K

C4: 100K



- ▶ An appropriate equation of state is used to calculate temperature
- ▶ The colder the initial temperature, the slower the growth and development of the jet
- ▶ For $T_{\text{initial}} < T^*$ (cases B4 & C4) the heat exchange does not affect the centerline temperature due to specific heat behavior in the near critical region



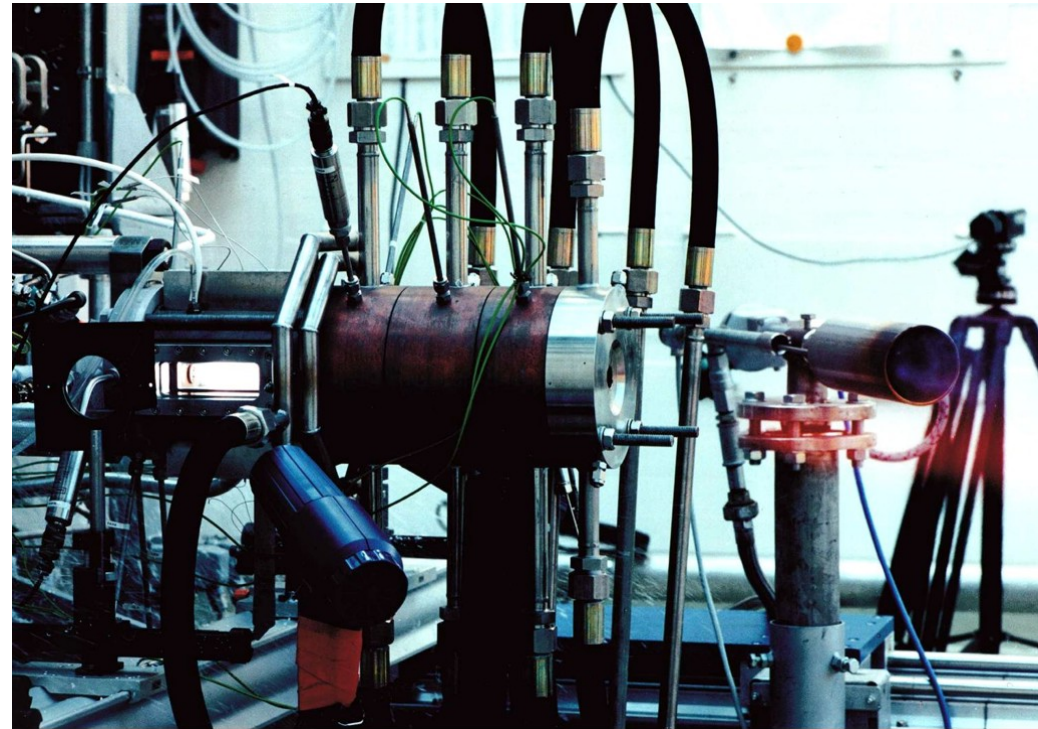
Combustion at representative pressure conditions

P8 test facility

- ▶ GH_2 , LH_2 supply
- ▶ CH_4 supply in preparation

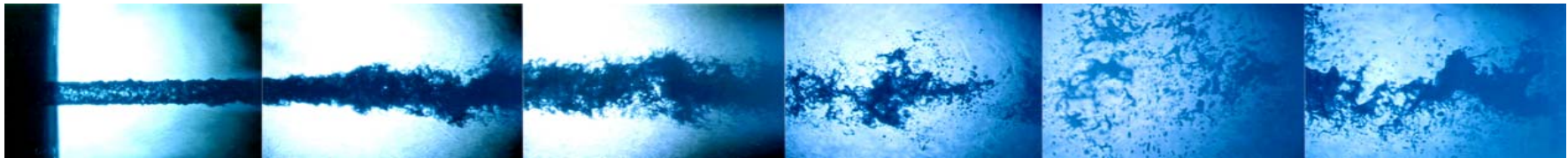
DLR combustion chamber "C"

- ▶ single injector head
- ▶ P_c up to 10 MPa, combustion at supercritical O_2 - and CH_4 -pressures
- ▶ optical access
 - shadowgraphy
 - OH-imaging
 - CARS

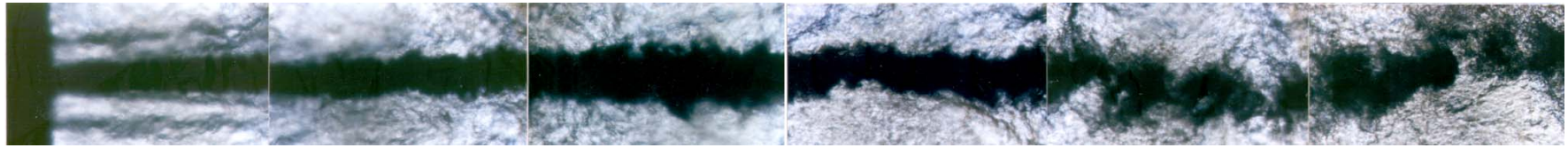


Combustion Studies LOX/H₂ at super critical pressure

LOX-jet disintegration:



(a) Subcritical Pressure, 1.5 MPa Combustion



(b) Supercritical Pressure, 10 MPa Combustion

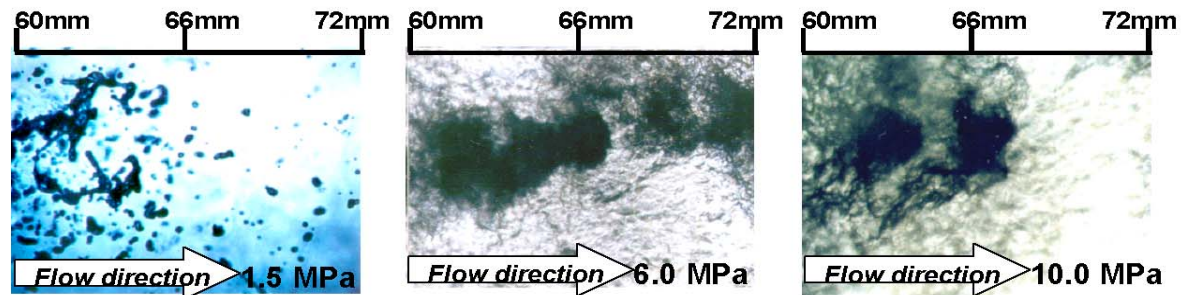
LOX-jet at subcritical (a) and supercritical (b) pressure conditions (from Mayer and Tamura)

▶ subcritical:

- disintegration into LOX-droplets

▶ supercritical:

- disintegration into O₂-clumps of larger size than typical liquid entities in subcritical case



Visualization of O₂-jet disintegration with varying chamber pressure (Mayer and Smith)

Conclusions

► injector scaling for reactive sprays

- non-dimensional parameters characterising cold flow conditions at injector exit not sufficient
- flame stabilization mechanism has significant influence on atomization process and droplet distribution in the flow
- coupling between atomization and combustion
- scaling has to take care for kinetics and transport properties

$\psi = \delta/h$ δ : laminar flame thickness, h : LOX-post thickness

(Juniper M., Candel S., Journal of Propulsion and Power, Vol. 19, No. 5, p. 332)

tests conditions should be as near as possible to representative conditions (propellants, pressure, ...) to get insight into relevant flame/spray interaction processes!

Perspectives

- ▶ influence of LOX-post wall thickness on lift-off behaviour
- ▶ GO_2/GCH_4 -ignition
- ▶ investigation of LOX/CH_4 spray combustion at supercritical pressure at P8 test facility (starting in July this year)